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### **Further Information**

Literature

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#### **Related Reactions**

Appel Reaction Staudinger Reaction Synthesis of esters

## Mitsunobu Reaction

HO H + R"COOH 
$$\xrightarrow{\text{DEAD}}$$
 H O  $\xrightarrow{\text{R"}}$  R"

HO H + N<sub>3</sub>H  $\xrightarrow{\text{DEAD}}$  H N<sub>3</sub>

R PPh<sub>3</sub> R'

The Mitsunobu Reaction allows the conversion of primary and secondary alcohols to esters, phenyl ethers, thioethers and various other compounds. The nucleophile employed should be acidic, since one of the reagents (DEAD, diethylazodicarboxylate) must be protonated during the course of the reaction to prevent from side reactions.

Suitable nitrogen nucleophiles include phthalimide or hydrogen azide; subsequent hydrolysis (in the case of using phthalimide, see Gabriel Synthesis) or selective reduction (in the case of azide formation, see Staudinger Reaction) makes the corresponding amines accessible.

#### **Mechanism of the Mitsunobu Reaction**

The triphenylphosphine combines with DEAD to generate a phosphonium intermediate that binds to the alcohol oxygen, activating it as a leaving group. Substitution by the carboxylate, mercaptyl, or other nucleophile completes the process.

$$O = (CO_2E)$$

$$O = (CO_2C)$$

$$O = (CO_2C)$$

$$O = (CO_2C)$$

$$O = (CO_2C)$$

The reaction proceeds with clean inversion, which makes the Mitsunobu Reaction with secondary alcohols a powerful method for the inversion of stereogenic centers in natural product synthesis.

## Side Reaction:

New protocols have been developed which allow better removal of side products and/or the conversion of more basic nucleophiles.

#### **Recent Literature**

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## Pronucleophile

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$$R-OH + H-Nu = \begin{array}{c} 1.2 \text{ eq.} \\ \hline 1.2 \text{ eq$$

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Organocatalytic Mitsunobu Reactions

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Synthesis of *N*-H vinylaziridines: a comparative study

B. Olofsson, R. Wijtmans, P. Somfai, Tetrahedron, 2002, 58, 5979-5982.

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Single-Step Process for the Reductive Deoxygenation of Unhindered Alcohols

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